Benzenoid Ring Functionalization of Indoles and Tryptophols via Combined **Directed** Ortho Metalation-Cross Coupling Methodology

Edward J. Griffen, David G. Roe, and Victor Snieckus*

Guelph-Waterloo Center for Graduate Work in Chemistry University of Waterloo, Waterloo, Ontario, Canada N2l 3G1

Received December 23, 1994

We report new strategies for the regiospecific construction of 4-, 5-, and 6- substituted indoles and tryptophols 2 by adaption of directed ortho metalation $(DoM)^{1,2}$ and transition metal catalyzed cross coupling^{1,3} reactions on N,N-diethylindole 5-O-carbamate 1 (Scheme 1). While the elaboration of the tryptamine side chain is readily effected by conventional means, versatile methods for the introduction of functionality into the benzene ring are lacking and usually involve classical electrophilic substitution or prior incorporation of substituents in de novo indole ring construction.⁴ Recent years have witnessed rapid advances in research on serotonin (5-hydroxytryptamine, 5-HT) and melatonin, neurotransmittors involved in a broad range of physiological processes.⁵ The remarkable heterogeneity of 5-HT receptors⁶ and the potential for the discovery of new pharmacological and therapeutic agents7 have served as powerful stimuli for the synthetic modification of indole and tryptamine derivatives.⁸ The preliminary results reported herein constitute the first direct and general methodology for the modification of the benzenoid moiety of indoles which may have value in the development of new synthetic neurotransmitters and in pharmacological studies related to 5-HT receptors.

On the basis of previous studies,^{4b,9,10} we anticipated that a bulky N-silicon group in 1 would prevent 2-deprotonation² and allow development of conditions for benzene ring metalation. In the event, treatment of $3^{11,12}$ (Scheme 2) under standard metalation conditions¹³ followed by quench with MeI gave $4a^{12}$ (Table 1) in

- (4) (a) The Chemistry of Indoles; Sundberg, R. J., Ed.; Academic: New York, 1970; pp 1 and 142. (b) See, however: Beswick, P. J.; Greenwood, C. S.; Mowlem, T. J.; Nechvatal, G.; Widdowson, D. A. Tetrahedron 1988, 44, 7325.
- (5) Miller, J. D. Ann. Rep. Med. Chem. 1992, 27, 11. McCall, R. B. Ann. Rep. Med. Chem. 1992, 27, 21.
- (6) Hoyer, D.; Clarke D. E.; Fozard, J. R.; Hartig, P. R.; Martin, G. R.; Mylecharane, E. J.; Saxena, P. R.; Humphrey, P. P. A. Pharmacol.

Rev. 1994, 46, 157. (7) Moon, M. W.; Morris, J. K.; Heier, R. F.; Chidester, C. D.; Hoffmann, W. E.; Piercey, M. F.; Althaus, J. S.; Voigtlander, P. F.; Evans, D. L.; Figur, L. M.; Lahti, R. A. J. Med. Chem. 1992, 35, 1076.

(8) For example, synthetic work has focused on the preparation of serotonin mimics exhibiting locked conformations or bearing additional functionality to probe receptor site selectivity: Swain, C. J.; Baker, R.; Kneen, C.; Herbert, R.; Moseley, J.; Saunders, J.; Seward, E. M.; Stevenson, G. I.; Beer, M.; Stanton, J.; Watling, K.; Ball, R. G. J. Med. Chem. 1992, 35, 1019. See also: Varie, D. L. Tetrahedron Lett. 1990, 31, 7583 and references cited therein. Varie, D. L. National Meeting of the American Chemical Society, Washington, D.C., Aug 1994; Abstract ORG-98

(9) Sundberg, R. J.; Russell, H. F. J. Org. Chem. 1973, 41, 3324.
(10) Iwao, M. Heterocycles 1993, 36, 29.

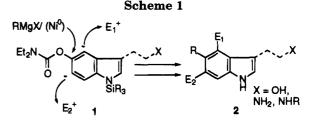


Table 1. Synthesis of 4-Substituted 5-O-Indole Carbamates

Entry	Starting Material	E ⁺ or Conditions ^a	Product (E)	Yid, %b
1	3	Mel	4a (Me)	99
2	3	DMF	4b (CHO) ^C	96
3	3	A	4c ^{AcO}	50
4	3	CICO ₂ Et	4d (CO2Et)	43
5	3	В	4e (NHAc)	47
6	3	С	4f (OAc) CONEt ₂	59
7	3	D	4g AcO	25
8	3	C2Cl6	4h (Cl)	90
9	3	I2	4 1 (I)	79
10	3	tBuSSO2Ph	4j (tBuS)	77
11	3	TMSC	4k (TMS)	90
12	4k	Mel	5 a (Me)	63
13	4k	12	5b (i)	63

^a A: (1) cyclohexanone, (2) AcOH/THF, (3) Ac₂O/Et₃N//DMAP (cat). B: (1) TsN₃, (2) EDTA, (3) Pd/C/H₂, (4) Ac₂O. C: (1) MgBr₂, (2) camphor sulphonyl oxaziridine, (3) AcOH, (4) Ac₂O/Et₃N. D: (1) ^tBuOOLi, (2) Ac₂O/Et₃N. ^b Yields represent purified (chromatographed) materials. ^c Quenched at -78 ^oC with AcOH/THF to avoid carbamate loss; see ref 13.

quantitative yield. Extension to other electrophiles indicates the generality of this process: functional carbon electrophiles at various oxidation states (entries 1-4), some of which are valuable for lysergic acid synthesis,¹⁴ NH_2^+ and OH^+ synthons (entries 5 and 6),¹⁵ halogens (entries 8 and 9), sulfur (entry 10),¹⁶ and silicon (entry 11) electrophiles are introduced in modest to excellent vields.

Following the concept of silicon protection for kinetically reactive anionic sites previously established in our

© 1995 American Chemical Society

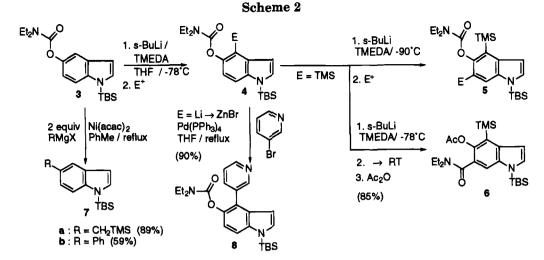
⁽¹⁾ Snieckus, V. Chem. Rev. **1990**, 90, 879. Rewcastle, G. W.; Katritzky, A. R. Adv. Heterocycl. Chem. **1993**, 56, 155.

⁽²⁾ For leading references to indole metalation chemistry, see: Gharpure, M.; Stoller, A.; Bellamy, F.; Firnau, G.; Snieckus, V. Synthesis 1991, 1079.

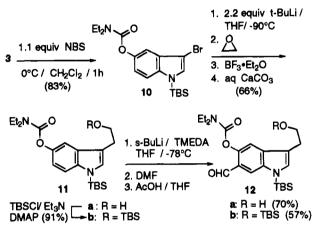
⁽³⁾ Quesnelle, C. A.; Familoni, O. B.; Snieckus, V. Synlett 1994, 349 and references cited therein.

⁽¹¹⁾ Compound 3 was routinely prepared on 20 gram scale as follows: commerical 3-methyl-4-nitrophenol was converted (1. NaH/ bMF; 2. Et₂NCOCl, 92%) into the corresponding carbamate which was subjected to the Leingruber-Batcho synthesis: 1. (MeO)₂CNMe₂/DMF; 2. 10% Pd/C/toluene/H2, 1 atm (Org. Synth. 1985, 63, 214) and silylation (NaH/TBSCl) to afford the product in 63% overall yield

without chromatography. (12) Salient NMR data. Compound 3: ¹H NMR δ 7.44 (d, J = 9.0, (12) Salient NMR data. Compound 3: ¹H NMR δ 7.44 (d, J = 9.0, 1H, H7), 7.34 (d, J = 2.4, 1H, H4), 6.92 (dd, J = 9.0, 2.4, 1H, H6). Compound **4a**: ¹H NMR δ 7.21 (d, J = 8.8, 1H, H7), 6.87 (d, J = 8.8, 1H, H6). Compound **12a**: ¹H NMR δ 7.35 (s, 1H, H4), 7.99 (s, 1H, H7). (13) Sibi, M. P.; Snieckus, V. J. Org. Chem. **1983**, 48, 1935. (14) Ninomiya, I.; Kiguchi, T. The Alkaloids; Brossi, A., Ed.; Academic Press: New York, 1990; Vol. 38; p 1.



Scheme 3



laboratories,¹⁷ the carbamate 4, E = TMS (Scheme 2), was subjected to the identical metalation conditions except at lower temperatures to avoid the anionic ortho-Fries rearrangement¹³ and quenched with electrophiles to give 6-substituted products **5a-b** (Table 1, entries 12 and 13), thus demonstrating the viability of selective 4and 6-substitution. In an attempt to hydroxylate 3, the anionic Fries was observed to give, after acylation, **4g** (entry 7). Under standard conditions¹³ followed by acylation, **4k** afforded **6**. Both products may be amenable to further metalation chemistry.

In order to demonstrate modern transition metal catalyzed cross coupling in which the indole derivative acts as the organometallic partner, the 4-lithio species 4, E = Li (Scheme 2), was transmetalated with ZnBr₂ following the excellent Negishi procedure¹⁸ and coupled with 3-bromopyridine to give compound 8 in very good yield.¹⁹ Successful cross coupling of iodinated products 4i and 5b (entries 9 and 13) may also be anticipated.

The recently established²⁰ Grignard reagent-aryl Ocarbamate cross coupling reaction was adapted to demonstrate viability of C-5 substitution. Thus, compound **3** was converted into **7a** and **7b** in good yields.²¹ In order to demonstrate synthetically advantageous metalation of tryptophol derivatives, the metalation-formylation of the 5-O-carbamate 11a (Scheme 3) and its side chain protected derivative 11b, prepared via 10 from the parent indole 3, was explored. Surprisingly, both 11a and 11b led, upon DMF quench, to the formation of 6-formylated products, 12a and 12b, respectively. This result which is perhaps associated with side-chain conformational effects²² suggests potential for direct modification of tryptophols and tryptamines.

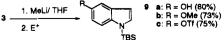
In conclusion, this work constitutes the first demonstration of a general DoM approach for functionalization of the benzenoid portion of indoles and tryptophols (1 -2) through the expediency of the 5-O-carbamate-directed metalaton group. Selective 4- and 6-substitution may be achieved via TMS protection and anionic ortho-Fries rearrangement tactics (4-6). Furthermore, organozinc and Grignard transition metal catalyzed cross coupling reactions may be adapted to derive powerful new protocols for C-C bond formation (7a,b, 8). The preliminary demonstration of tryptophol metalation chemistry (12a,b) also anticipates a broader perspective. The utility of this methodology, separate or combined, for the modification of indole derivatives in total synthesis endeavors or in pharmacological studies may be anticipated. Extension of these studies are in progress.

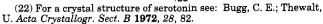
Acknowledgment. The support of Glaxo Canada, Inc., and the government of Ontario under the URIF program (Grant) was decisive in achievement of this research. Stimulating discussions with Dr. E. Collington and Dr. P. North, Glaxo, Greenford, U.K. are gratefully acknowledged.

Supplementary Material Available: General experimental and characterization data (27 pages).

JO942172I

⁽²¹⁾ In order to demonstrate deprotection of the 5-hydroxy function, the carbamate **3** was treated with MeLi and either worked up hydrolytically or treated with MeI or 2-pyN(Tf)₂ to give **9a**, **9b**, and **9c**, respectively. *Inter alia*, this may allow the preparation of 5-hydroxytryptamine analogues.





^{(15) 4-}Hydroxy- and 4-aminoindoles are structural components of psylocibin and teleocidin alkaloids; see: Nakatsuka, S-i.; Masuda, T.; Asano, O.; Teramae, T.; Goto, T. *Tetrahedron Lett.* **1986**, 27, 4327 and references cited therein.

⁽¹⁶⁾ For the Li to Sn transmetalation method for s-electrophile introduction, see: Dickens, M. J.; Mowlem, T. J.; Widdowson, D. A.; Slawin, A. M. Z.; Williams, D. J. J. Chem. Soc., Perkin Trans. 1 **1992**, 323.

⁽¹⁷⁾ Mills, R. J.; Taylor, N. J.; Snieckus, V. J. Org. Chem. **1989**, 54, 4372.

⁽¹⁸⁾ Negishi, E.-i.; King, A. O.; Okukado, N. J. Org. Chem. 1977, 42, 1821.

⁽¹⁹⁾ Use of modified Suzuki conditions ((1) $B(OMe)_3$; (2) $Pd(PPh_3)_{4/2}$ M $K_2CO_3/DME/reflux)$ on 4, E = Li, afforded the analogous crosscoupled product with loss of TBS in 19% yield. For other low yield couplings of pyridyl boron derivatives with 4-iodo- and 4-thallioindoles, see: Somei, M.; Amari, H.; Makita, Y. Chem. Pharm. Bull. Jpn. 1986, 34, 3971.

⁽²⁰⁾ Sengupta, S.; Snieckus, V. J. Org. Chem. 1990, 55, 5680.